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# Third Semester M.Sc. Degree Examination, February 2021

## **Polymer Chemistry**

## PC 232 : ORGANIC CHEMISTRY II

## (2014 to 2017 Admission)

Time : 3 Hours

Max. Marks: 75

### SECTION - A

Answer any two among (a), (b) and (c) from each question. Each sub question carries 2 marks.

- 1. (a) Explain "antiaromaticity" with examples.
  - (b) Explain very briefly the stereo-aspects of Diels-Alder reaction.
  - (c) Cyclopropenone and cycloheptatrienone are relatively stable compounds whereas, cyclopentadienone is unstable. Why?
- 2. (a) What is Barton reaction? Give an example.
  - (b) Write any two applications of photoreactions in industrial synthesis.
  - (c) What are Norrish reactions? Give one example each.
- 3. (a) How is enamine prepared? Show one of its applications in organic synthesis.
  - (b) As a reducing agent, what are the advantages of sodium borohydride over lithium aluminium hydride?
  - (c) Explain the stereo selective hydroxylation of an alkene with OsO<sub>4</sub>.

- (a) Explain hydrogen bonding stretching frequency and how will you differentiate inter and intra molecular hydrogen bonding in IR spectra.
  - (b) Give the structure of the compound  $C_3H_9N$  with one <sup>1</sup>H and one <sup>13</sup>C-NMR peaks.
  - (c) How can you confirm the presence of bromine in an organic compound using mass spectroscopy?
- (a) Write down the structure of the product/s in the reaction of 2-cyclohexanone with (CH<sub>3</sub>)<sub>2</sub>CuLi and then with H<sub>3</sub>O<sup>+</sup>.
  - (b) Write the structure of any one nucleotide with proper numbering to different positions.
  - (c) Explain very briefly the role of protecting groups in nucleotide synthesis by giving suitable examples.

 $(10 \times 2 = 20 \text{ Marks})$ 

### SECTION - B

Answer either (a) or (b) from each question. Each sub question carries 5 marks.

- 6. (a) Write short notes on the aromaticity of
  - (i) fullerenes and
  - (ii) metallocenes
  - (b) Illustrate the stereochemical aspects and the mechanism of Diels Alder reaction and Cope rearrangement
- 7. (a) Discuss the photoreactions of vitamin D.
  - (b) Write notes on:

11.

- (i) di-pi methane rearrangement.
- (ii) Singlet and triplet state reactivities in photoreactions.

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- (a) What is Mannich base? How it is prepared? Give two applications of Mannich reaction in organic synthesis.
  - (b) Illustrate that Peterson olefination reaction can be used for preparing either *cis* or *trans* alkenes from the same  $\beta$  hydroxysilane.
- 9. (a) Explain the changes produced in the auxochromic red shift of the following:
  - (i) basifying C<sub>6</sub>H<sub>5</sub>COOH
  - (ii) acidifying C<sub>6</sub>H<sub>5</sub>NH<sub>2</sub>
  - (b) Explain the following in NMR spectra.
    - (i) acetylenic H's are at a higher field than vinylic H's
    - (ii) the  $\delta$  value of an aromatic H (6- 8.5 ppm) is higher than a vinylic H (4.6-5.9)
- 10. (a) Describe the synthesis of any two nucleic acid bases.
  - (b) Explain the automated oligonucleotide synthesis by phosphoramidite method.

 $(5 \times 5 = 25 \text{ Marks})$ 

#### SECTION - C

Answer any three questions. Each question carries 10 marks.

- 11. What are pericyclic reactions? Discuss the classification of pericyclic reactions giving suitable examples.
- 12. Write a detailed note on synthetically useful photoreactions of carbonyl compounds, enes and dienes.
- 13. Write short notes on the following name reactions (including mechanism) in organic synthesis with two applications each.
  - (a) Mc Murray reaction
  - (b) Robinson ring annulation
  - (c) Shapiro reaction
  - (d) Reforxmatsky reaction

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- 14.
- Discuss the basic fragmentation types and factors influencing fragmentation pattern in carbonyl compounds using appropriate examples. 15. Discuss the applications of the following reagents in organic synthesis

  - (b) DDQ
  - (c) DIBAL
  - $(n-C_4H_9)_3SnH$ (d)

(3 × 10 = 30 Marks)